



# Assessment of methods for measuring the membrane-water partition ratio ( $K_{MW}$ ) for surfactants

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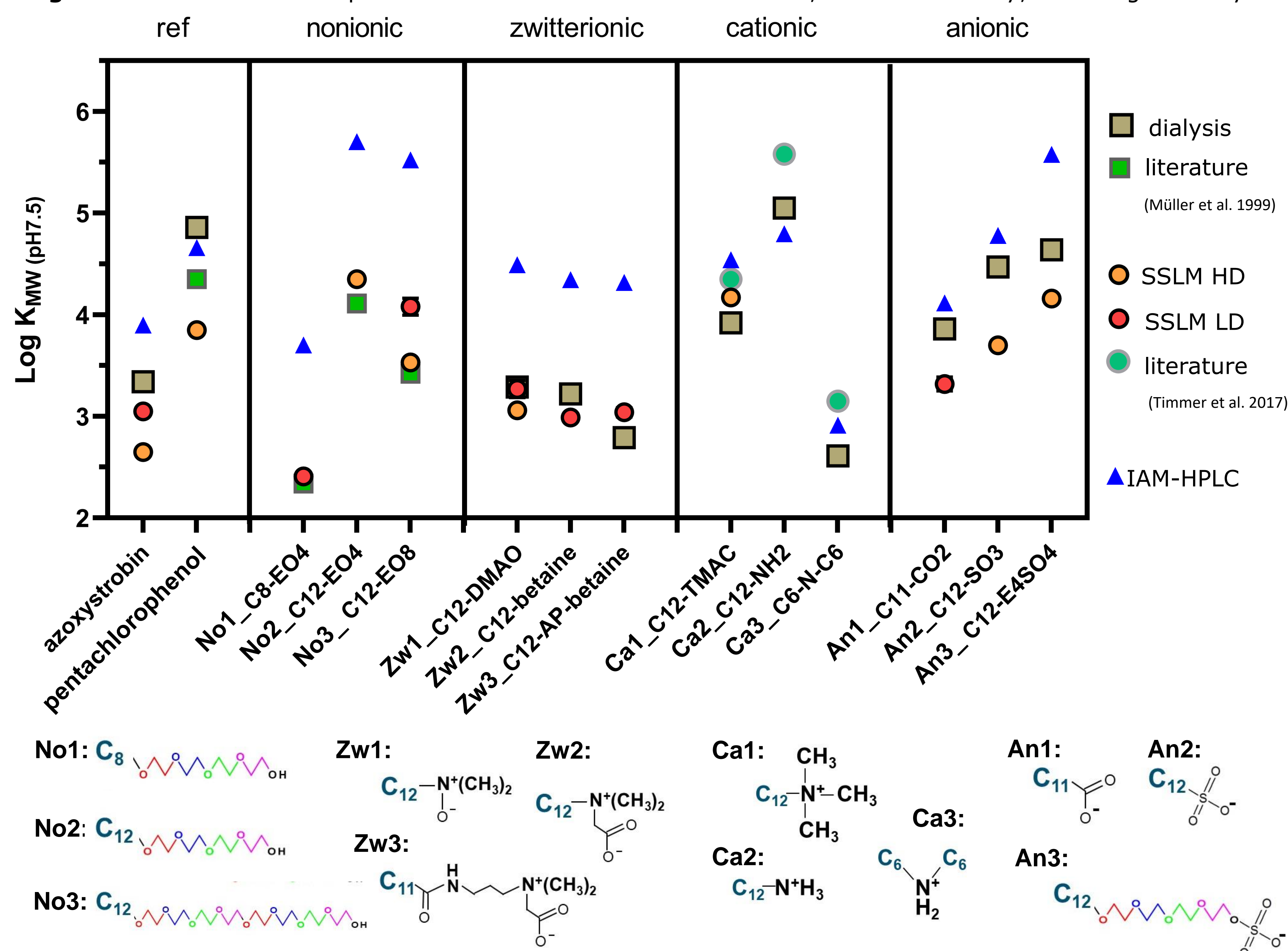
## Background & Objective

The Environment and Health – Risk Assessment & Management (ERASM) is a joint research platform of the European Detergents and Surfactants Industries. The ERASM ‘Membrane Water Partitioning of Surfactants’ project aims to evaluate the alignment between 3 experimental and 3 computational methods to derive the **phospholipid membrane-water partition ratio ( $K_{MW}$ )**, covering 4 surfactant types.  $K_{MW}$  previously has been shown to provide a New Approach Methodology (NAMs - 3R principle) to assess the bioconcentration factor (BCF) when applied with *in vitro* biotransformation data as part of a tiered approach (Droge et al. 2021). Here we discuss results from 3 experimental methods: liposome vesicles in dialysis systems, Solid Supported Lipid Membranes (SSLM: bilayers covering silica), and immobilized artificial membrane chromatography (IAM-HPLC: monolayer coated silica). For computational methods, see Poster Corner TUE 3.12 Kearney et al.

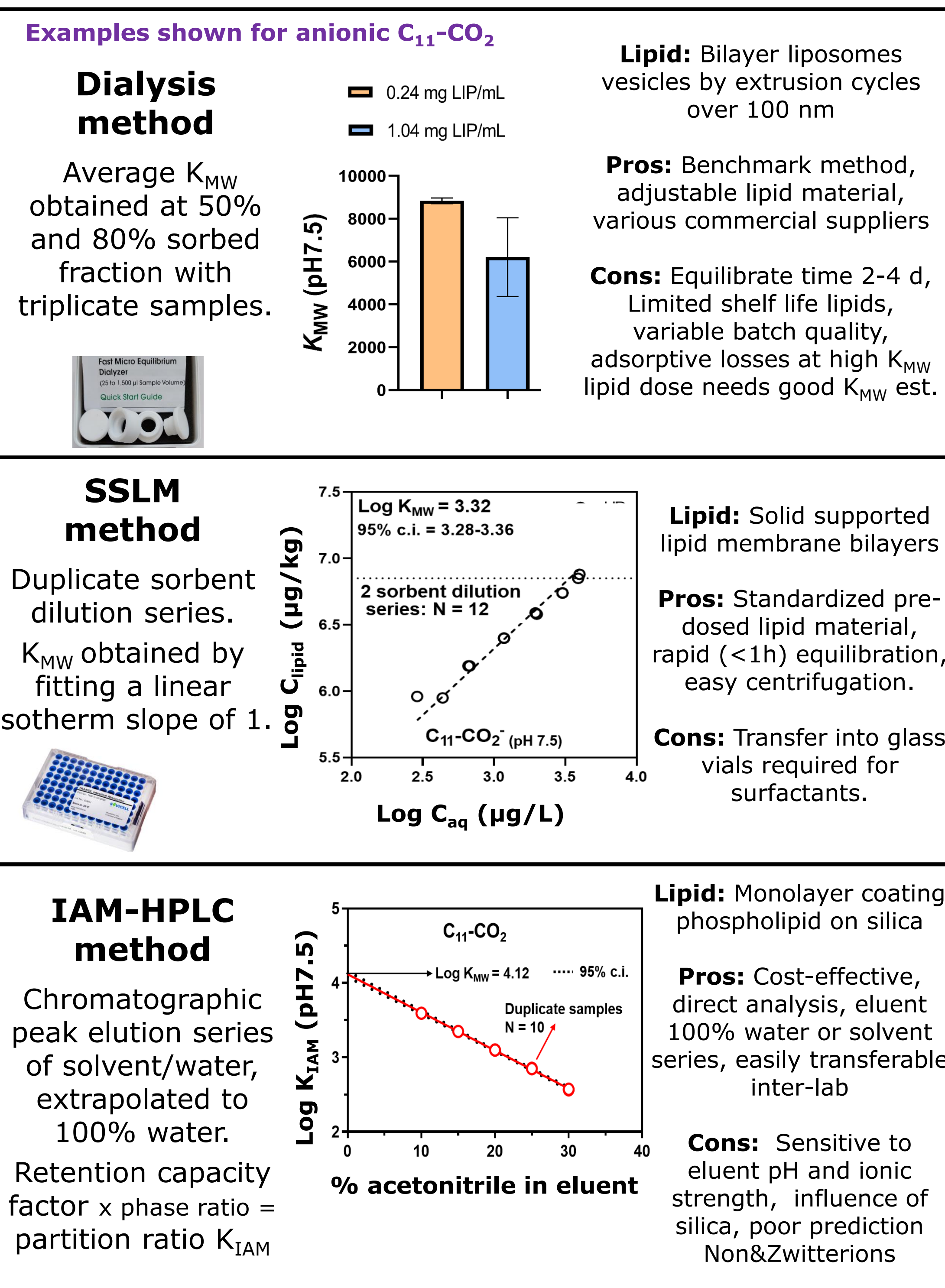
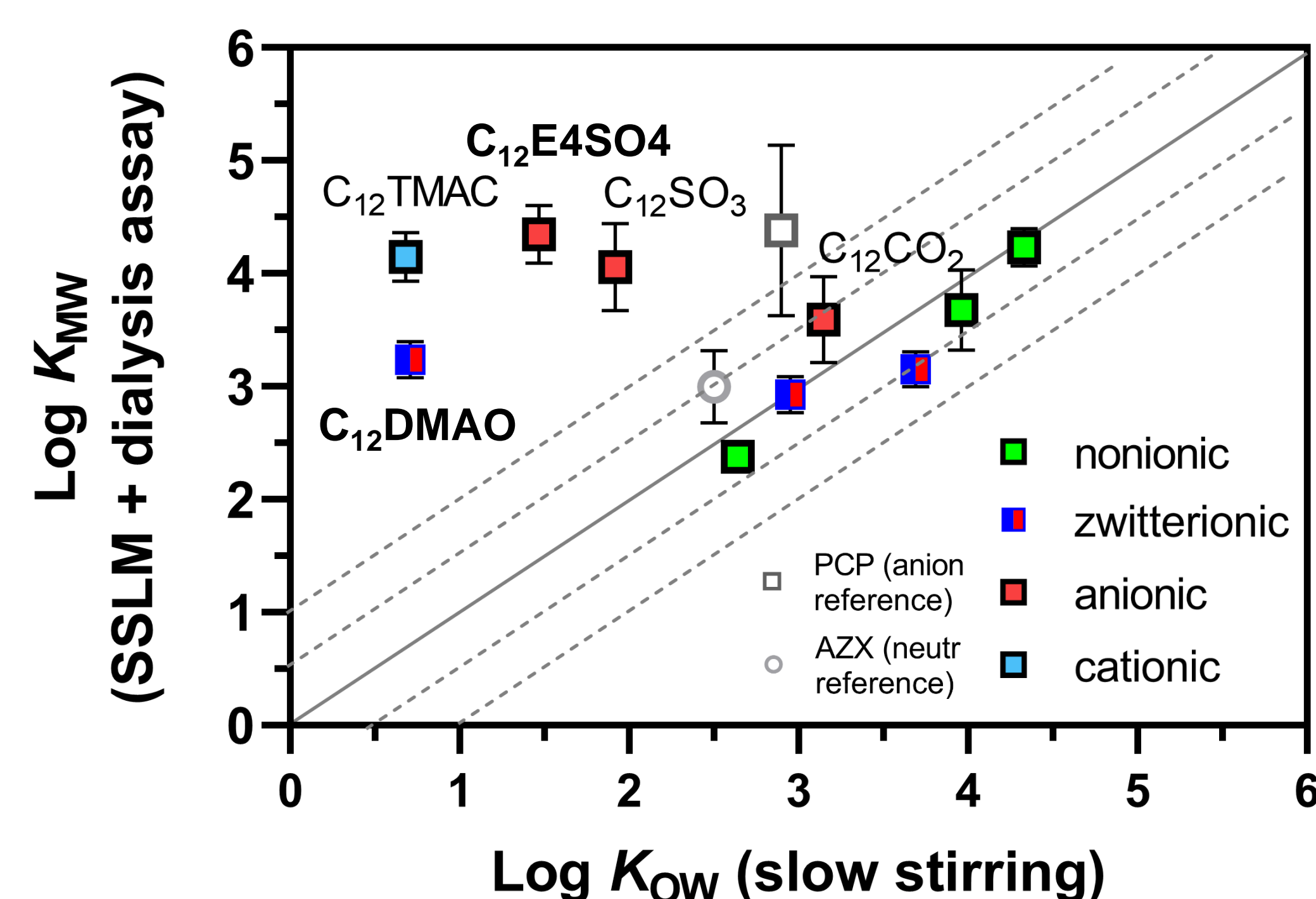
## Results

- Even for non-surfactant chemicals,  $K_{MW}$  range  $\sim 1$  log units for the 3 methods (**Figure 1**).
- There is reasonable consistency between the  $K_{MW}$  by SSLM and dialysis. The high density (HD) and low density (LD) SSLM sorbent dilution series deviated up to a factor 3 in  $K_{MW}$ . Care is needed in selecting the right density (**Figure 1**).
- $K_{MW}$  by IAM-HPLC are  $>10\times$  higher than SSLM for zwitterions, nonionics and C12-E4SO<sub>4</sub>, but lower for some cations (**Figure 1**). The ethoxylate chains (coloured units in molecules) cause strong overestimation for IAM-HPLC, no explanation yet for zwitterions.
- $K_{OW}$  is predictive of  $K_{MW}$  for nonionic surfactants, but  $K_{OW}$  may strongly underestimate the favourable interactions with phospholipids for ionic surfactants. Biotransformation data maybe required for some surfactants to determine relevant BCF values. For most ionic surfactants,  $K_{MW} \gg K_{OW}$  (**Figure 2**).
- Experimental feasibility for the experimental methods is limited to  $\log K_{MW} < 6$ . In a proposed Tier 1 baseline BCF assessment (no biotransformation, 1% phospholipid fraction as key sorptive phase, (Droge et al. 2021))  $\log K_{MW} 5 \equiv 1000$  L/kg for ionic surfactants. Validated computational approaches to predict  $K_{MW}$  are thus needed to extend the chemical domain for challenging (technical) surfactants.

**Figure 1:** Membrane-water partition ratios from 3 different methods, LD = low density; HD = high density



**Figure 2:** Average membrane-water partition ratios (SSLM and dialysis) plotted against the most reliable  $K_{OW}$  values using the slow stirring method (Hodges et al. 2019).



## Conclusions & Outlook

- We completed a systematic  $K_{MW}$  data matrix (**Figure 1**) to inform risk managers which type of membrane-water assay has a high confidence level, for which chemical applicability domain, and for further work towards regulatory acceptance.
- Commission Regulation (EU) Amendment 2021/979 states that: “an experimental bioaccumulation study cannot be waived on the basis of low  $K_{OW}$  alone, if the substance is surface active or ionisable at environmental pH (4–9).” This is confirmed for several ionic surfactants (**Figure 2**).
- The  $K_{MW}$  can be most effectively measured for all surfactant types using the SSLM method.  $K_{MW}$  determined by the SSLM method demonstrates good consistency with  $K_{MW}$  determined by the benchmark liposome dialysis method ( $< \text{factor } 3$ ). IAM-HPLC can roughly indicate  $K_{MW}$  for anionics and cationics.
- A ring test  $K_{MW}$  study is being considered to support possible development of OECD test guideline for  $K_{MW}$  analysis.
- Study results support earlier assessments (Hodges et al. 2019, Droge et al. 2021) that the  $K_{MW}$  is a biologically relevant & methodologically alternative to  $K_{OW}$  for ionizable surface-active chemicals.