

Coarse-Grained Molecular Dynamics Simulations of Passive Partitioning of Ionic Surfactants into Cell Membranes

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Objectives

- The Environment and Health – Risk Assessment & Management (ERASM) is a joint research platform of the European Detergents and Surfactants Industries. The ERASM ‘Membrane Water Partitioning of Surfactants’ project aims to evaluate the alignment between 3 experimental and 3 computational methods to measure the phospholipid membrane-water partition ratio (K_{mw}) for 12 surfactant structures, covering 4 surfactant types. This poster focuses on computational methods.
- Previously our group developed an automatic coarse-graining script to allow rapid setup of membrane-water partitioning simulations using molecular dynamics [1].
- We aim to derive best practice for use of the Martini coarse-grained force field for simulation of charged surfactants, and ultimately to benchmark it and other computational methods.

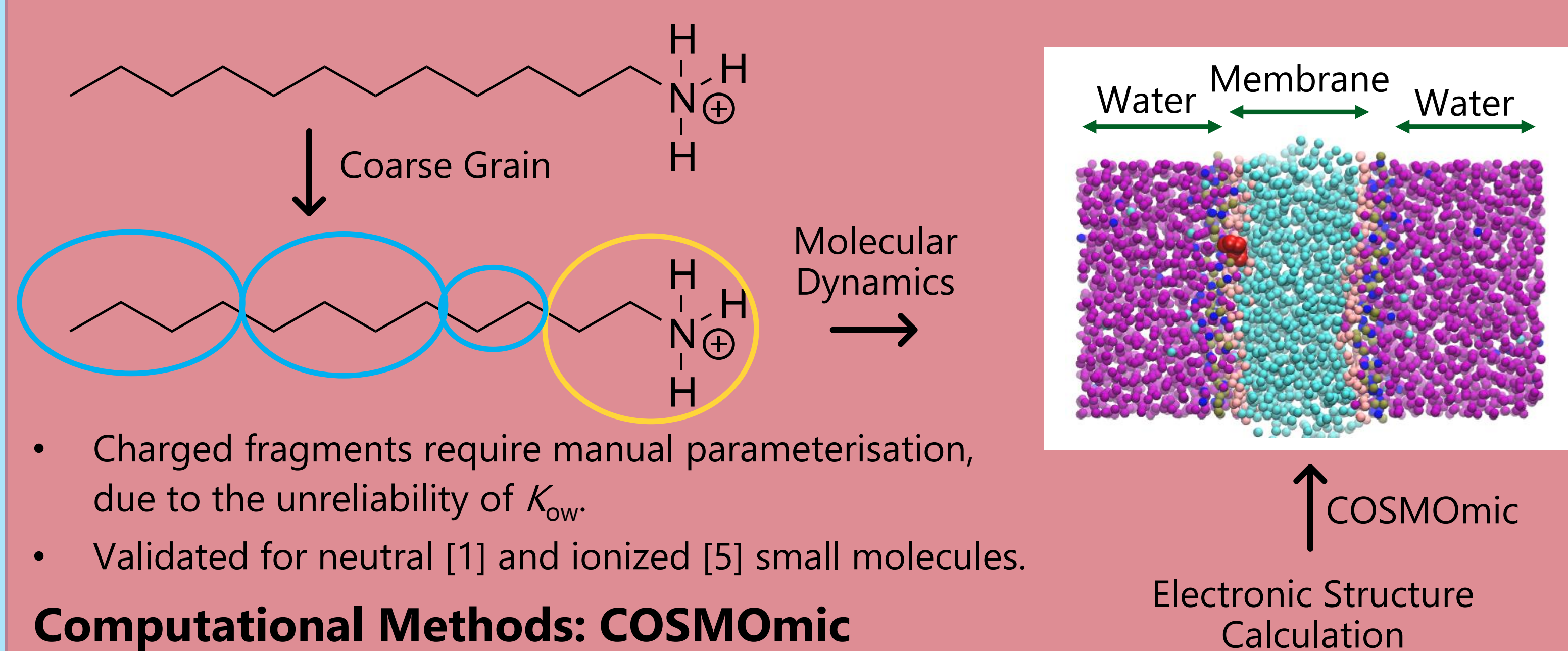
Background

- Membrane-water partitioning K_{mw} (or D_{mw}) is a key metric for baseline toxicity (narcosis) and bioaccumulation.
- Octanol-Water Partitioning, K_{ow} , has well-publicised [2] deficiencies for charged surfactants; $\log K_{mw}$ is a more representative alternative.
- Surfactants are an experimentally challenging species to study i.e. difficulties studying mixtures, issues with compound solubility or adhesion to glassware; simulation can overcome these problems.
- Goal: High-throughput models for K_{mw} simulation of surfactants.
- Membrane simulations can then be performed and are related to experiment via calculation of the Gibbs energy, ΔG :

$$K_{MW} = \frac{[\text{Solute}]_{\text{Membrane}}}{[\text{Solute}]_{\text{Water}}} = e^{-\Delta G/RT}$$

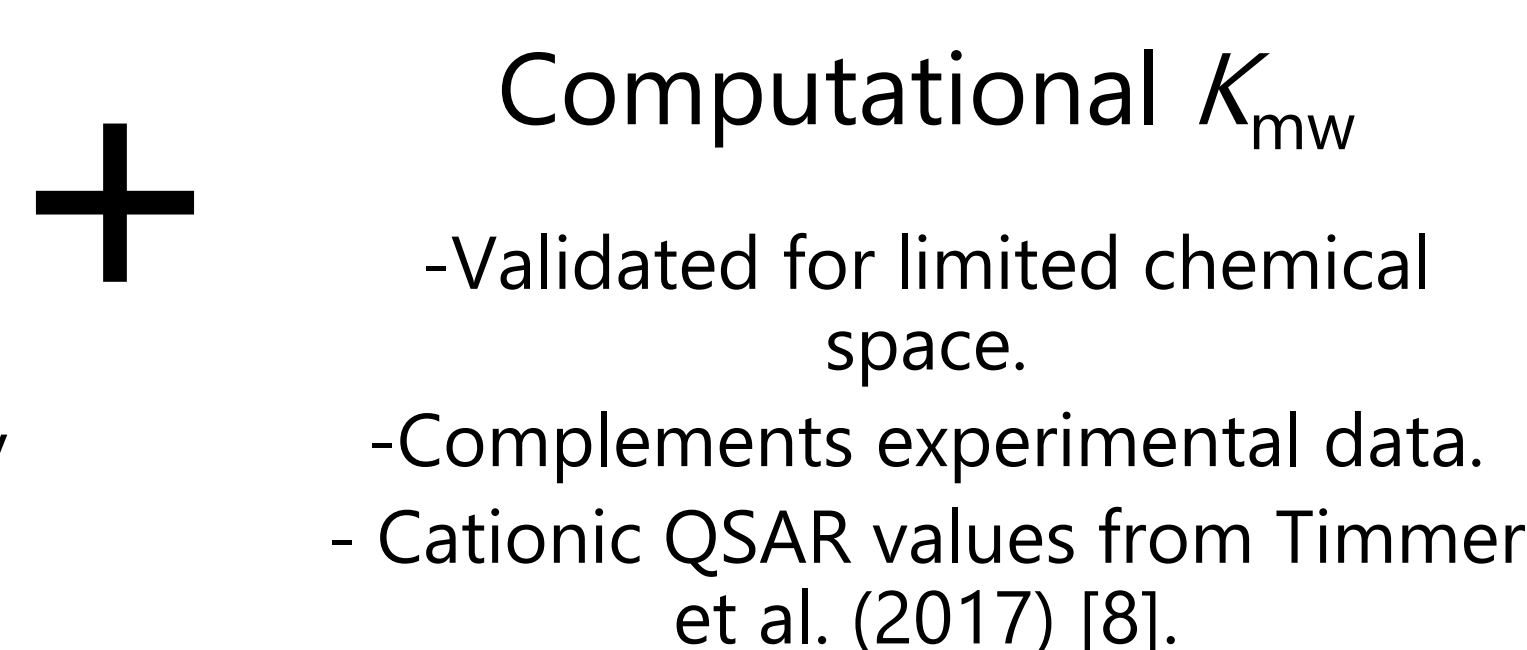
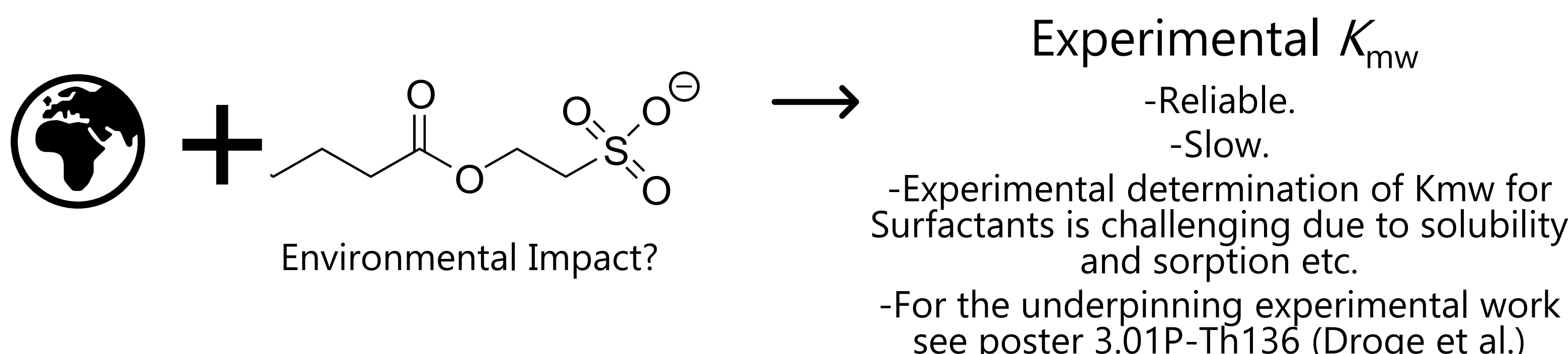
Computational Methods: Molecular Dynamics Simulation

- cg_param is a python script to convert a SMILES code into a coarse-grained structure, ready for simulation in a coarse-grained membrane.
- The script uses two main parts:
 - A graph-based spectral mapping algorithm to break large molecules up into roughly four-atom beads while preserving symmetry.
 - ALOGPS [3] a web-based neural network to generate $\log K_{ow}$ values for the fragments, allowing parameterisation into the Martini force field [4].



Computational Methods: COSMOmic

- Widely used proprietary software to calculate partition coefficients [6].
- Combines quantum chemical calculations of atomic orbitals with a continuum solvation model, thus avoid simulating many water molecules.
- Simulations of charged species use the membrane potential developed by Bitterman et al [7].
- Issues in simulation of cationics were uncovered: data below uses literature values (Timmer et al. values [8]).



3 Methods

QSPR

- Quantitative Structure-Property Relationship.
- Sum atomic contributions.
- Site of future work.

Molecular Dynamics

- Classical simulation using the Martini force field.
- Simulate movement of molecule through cell membrane, as an ongoing trajectory.
- Coarse-Grained to allow rapid simulation.

COSMOmic

- Electronic Structure Method.
- Calculate atomic orbital contributions.

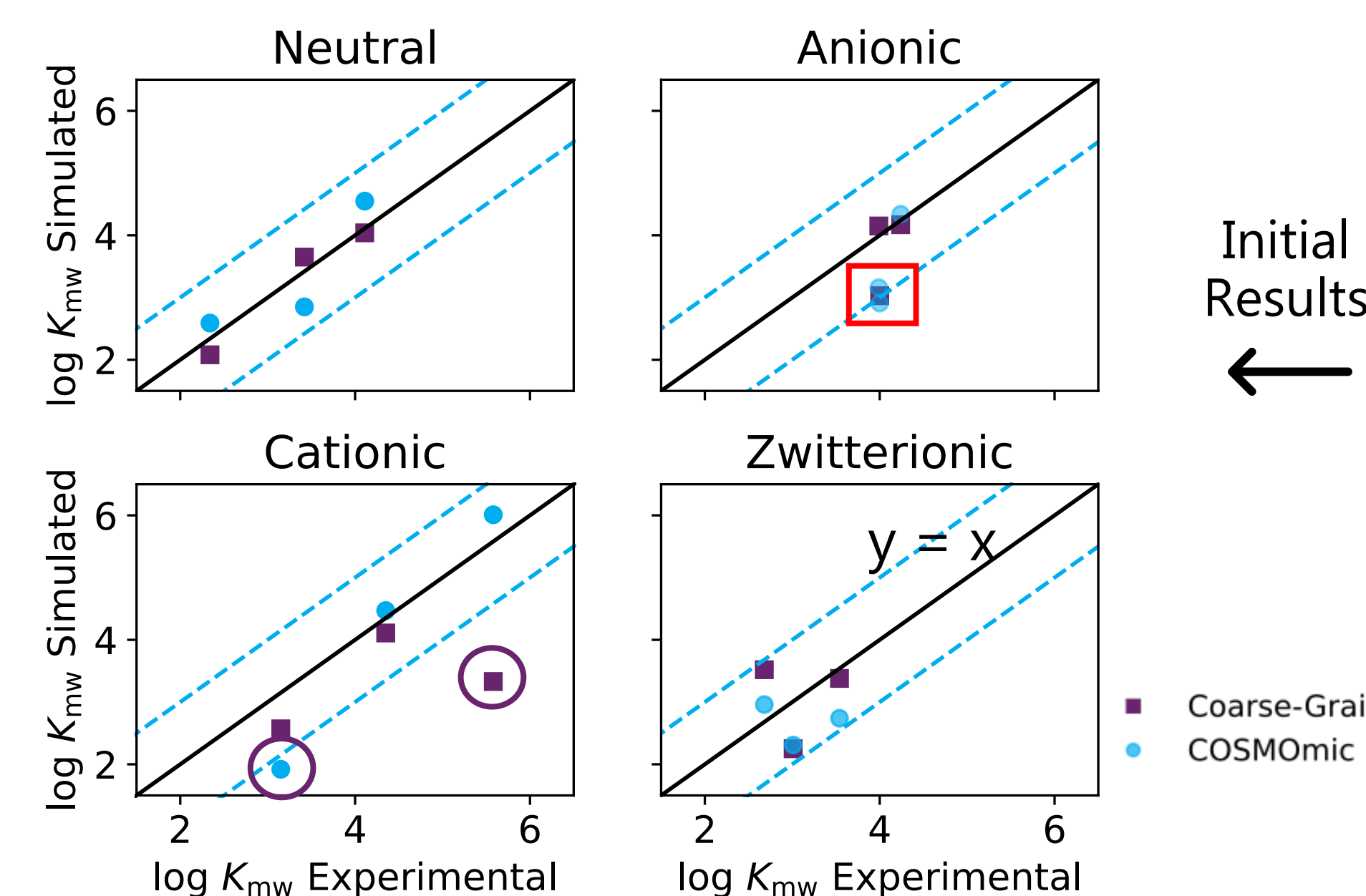
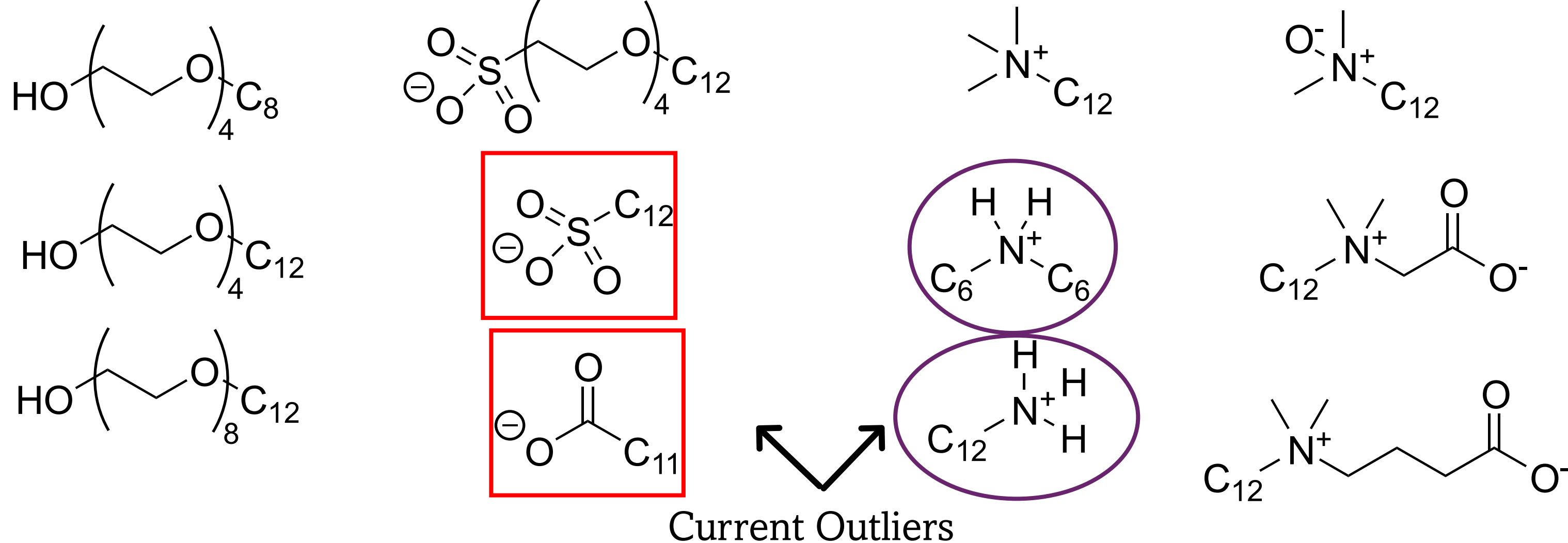
ERASM Project Surfactants

Neutral

Anionic

Cationic

Zwitterionic



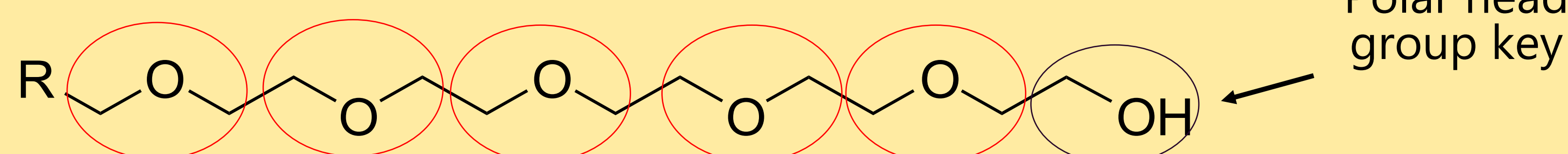
Results

- As part of ERASM, 12 representative surfactants were picked and subjected to computational and experimental study with the goal of establishing best practice when applying K_{mw} for surfactants.
- Results comparing simulated and experimental ERASM K_{mw} values are in the graph above right.

- Good performances for 10/12 molecules with both methods: Prediction for all surfactants within 1 log unit achieved with values from both methods.
- Note cationic COSMOmic values are sourced from Timmer et al. (2017) due to issues with recent simulations; cationics are especially sensitive to membrane potential values [8].
- Standard deviation of replica coarse-grained simulations: ~0.15.
- Molecular Dynamics Outliers: C12CO2 and C12NH3 – variably ionizable species.
- COSMOmic Outliers: C12SO3 and Dihexylamine.

Molecular Dynamics Development

- Custom parameterization for ether functional group were required.
- Ethers are a known issue in simulation [9]: individual ethers are sparingly hydrophilic and well-captured, but several ethers combine to form hydrophilic section of the molecule.
- Default coarse-grained models fail to capture this effect, but custom parameterisation targeting polyethers allows capture of this effect for the ERASM K_{mw} values.



Future work:

- Improve coarse-grained and COSMOmic parameterisations to improve results, and compare with alternative computational methods, e.g. pp-LFER [10].
- Full compilation and evaluation of K_{mw} results with experimental values.

Bibliography

- [1] T. Potter et al., J. Chem. Theory Comput., (2021), 17, 5791 (Initial release of cg_param, validated for a range of small molecules).
- [2] G. Hodges, et al., Environ. Sci. Europe, (2019), 3, 1-18 (Compares K_{ow} predictions for a set of surfactants; finds deficiencies for ionics and recommends K_{mw} as an alternative).
- [3] I. Tetko and P. Bruneau, J. Pharm. Sci., (2004), 93, 3103-3110 (Neural network trained to generate K_{ow} values).
- [4] P. Souza et al., Nat. Meth., (2021), 18 382-388 (Launch of Martini 3 force field).
- [5] T. Potter et al., Environ. Sci.: Processes Impacts, (2023), 25, 1082-1093 (Expansion of model to new Martini 3 force field, validated for small charged ions in membranes with cholesterol).
- [6] A. Klamt et al., J. Phys. Chem. B, (2008), 112, 12148-12157 (Adaptation of COSMO method for membranes, producing COSMOmic).
- [7] K. Bitterman et al., J. Phys. Chem. B, (2014), 118, 14833-14832 (Derivation of membrane potential to apply COSMOmic to ionic species)
- [8] N. Timmer and S. Droge, Environ. Sci. Technol., (2017), 51, 2890-2898 (COSMOmic cationic values).
- [9] K. Prasinok M. Wilson, Phys. Chem. Chem. Phys., (2013), 15, 17093-17104 (Custom coarse-grained parameterization of polyethylene glycol)
- [10] S. Endo et al., Anal. Chem., (2011), 8, 1394-1400 (Derivation of pp-LFER QSAR for K_{mw}).